## Synthesis of $(\pm)$ -Atractylon

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WE recently reported a new method<sup>1</sup> for synthesis of the furan moiety. In this paper this method applied for the synthesis of a furanois sesquiterpene, atractylon (I). Atractylon\* was isolated from Atractylis ovata Thunb.<sup>2</sup> and Atractylodes japonica Koizumi,3 and its structure was established as (I) by Yoshioka and coworkers.<sup>4</sup> We have used as the starting material 2,3,4,6,7,8-hexahydronaphthalene-2-one (III) which was derived<sup>5</sup> from (II).

According to Nagata's method,<sup>6</sup> (III) was treated with hydrogen cyanide and triethylaluminium in dry tetrahydrofuran to give two monocyano-ketones, (IV), m.p. 105-106° [vmax 2225, 1680, and 1630 cm.<sup>-1</sup>,  $\lambda_{\rm max}$  235 m $\mu$  ( $\epsilon$  16,300)] and (V), m.p. 95-96° [vmax 2225, 1675, and 1630 cm.<sup>-1</sup>,  $\lambda_{max}$  235 m $\mu$  ( $\epsilon$  16,250)] in 30%

yield. As the ratio of the optical density of a band at 2225 cm.<sup>-1</sup> in the infrared spectrum due to the cyano-group in (IV) to that in (V) is about 1.6:1, (IV) has an equatorial cyano-group and (V) an axial cyano-group.<sup>7</sup>

Repetition of hydrocyanation of (IV) afforded only one dicyano-ketone (VI), m.p. 172-174°  $(v_{max} 2225 \text{ and } 1725 \text{ cm.}^{-1})$  in 85% yield. This fact indicates that compound (VI) is a transfused decanone derivative, according to Nagata's studies<sup>6</sup> on the stereochemistry of cyanoketones.

The ethylene ketal (VII), m.p. 136-137°, of (VI) was hydrogenated with Raney nickel in ethanol-aqueous ammonia to give in a quantitative yield a monoaminomethyl derivative, which was further reduced with sodium borohydride in

\* The authors are very much indebted to Associate Prof. H. Hikino (Tohoku University) for his kindness in identifying atractylon.

<sup>1</sup> H. Minato and T. Nagasaki, Chem. and Ind., 1965, 899.

<sup>2</sup> S. Takagi and G. Hongo, J. Pharm. Soc. Japan, 1925, 44, 539.

<sup>3</sup> I. Yoshioka, H. Hikino, and Y. Hikino, unpublished data.

<sup>4</sup> H. Hikino, Y. Hikino, and I. Yoshioka, Chem. and Pharm. Bull. (Japan), 1962, 10, 641; 1964, 12, 755.
<sup>5</sup> N. N. Gaidamovich and I. V. Torgov, Bull. Acad. Sci. U.S.S.R., 1961, 1682.
<sup>6</sup> W. Nagata, M. Yoshioka, and S. Hirai, Tetrahedron Letters, 1962, 461; W. Nagata, "Collection of Lectures Commemorating the Inauguration of the New Shionogi Research Laboratory Building," 1961-1962, p. 248; W. Nagata, I. Kikkawa, and M. Fujimoto, Chem. and Pharm. Bull. (Japan), 1963, 11, 226.

<sup>7</sup> W. Nagata, M. Yoshioka, N. Narisada, and H. Watanabe, *Tetrahedron Letters*, 1964, 3133.



formaldehyde and methanol to give a dimethylamino-compound (VIII), m.p.  $131-132^{\circ}$  ( $\nu_{max}$ 2225 cm.<sup>-1</sup>) in 75% yield. Aluminium di-isobutyl hydride reduction<sup>8</sup> of (VIII), followed by modified Huang Minlon reduction<sup>9</sup> furnished (IX), m.p. 42-43°, in 70% yield. The N-oxide (X), m.p. 97-98°, which was obtained on treatment with *m*-chloroperbenzoic acid, on heating at 120–130°, followed by treatment with 80% acetic acid gave compound (XI), b.p.  $90^{\circ}/5$  mm. ( $\nu_{max}$  1710, 1640, and 890 cm.<sup>-1</sup>) in 55% yield.

The pyrrolidine enamine of (XI) was alkylated with ethyl  $\alpha$ -bromopropionate and hydrolyzed with 10% potassium carbonate in methanol to

<sup>8</sup> L. I. Zaharkin and I. M. Khorlina, Doklady Akad. Nauk. S.S.S.R., 1957, 116, 422.

<sup>&</sup>lt;sup>9</sup> W. Nagata and H. Itazaki, Chem. and Ind., 1964, 1194.

give an oily ketocarboxylic acid (XII) ( $v_{max}$  1710, 1640, and 890 cm.<sup>-1</sup>). This compound (XII) was converted into the  $\alpha\beta$ -unsaturated  $\gamma$ -lactone† (XIII), m.p. 102—103° ( $v_{max}$  1738, 1685, 1645 and 894 cm.<sup>-1</sup>) on heating with acetic anhydride and sodium acetate. The  $\gamma$ -lactone function of (XIII) was reduced to the furano-derivative by our new method, that is, on reduction with aluminium di-isobutyl hydride in tetrahydrofuran, (XIII) afforded racemic compound (I), b.p. 105—

110°/1.5 mm. (bath temp.) ( $\nu_{max}$  3090, 1645, 1565, 1136, 890, and 848 cm.<sup>-1</sup>; maleic anhydride adduct, m.p. 93—95°), which was shown to be identical with atractylon by comparison of the infrared spectrum.

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 $\dagger$  This compound was shown to be identical with the autoxidation product<sup>4</sup> of atractylon (I) by comparison of the i.r. spectrum. The authors wish to thank Prof. I. Yoshioka (Osaka University) for his courtesy of supplying the sample of the natural compound.